

Fig. 1 Mount Everest 1968 VS. 2007

Global Warming

At what point does atmospheric greenhouse gas release become unethical?

Luke Good & Gladys Lopez

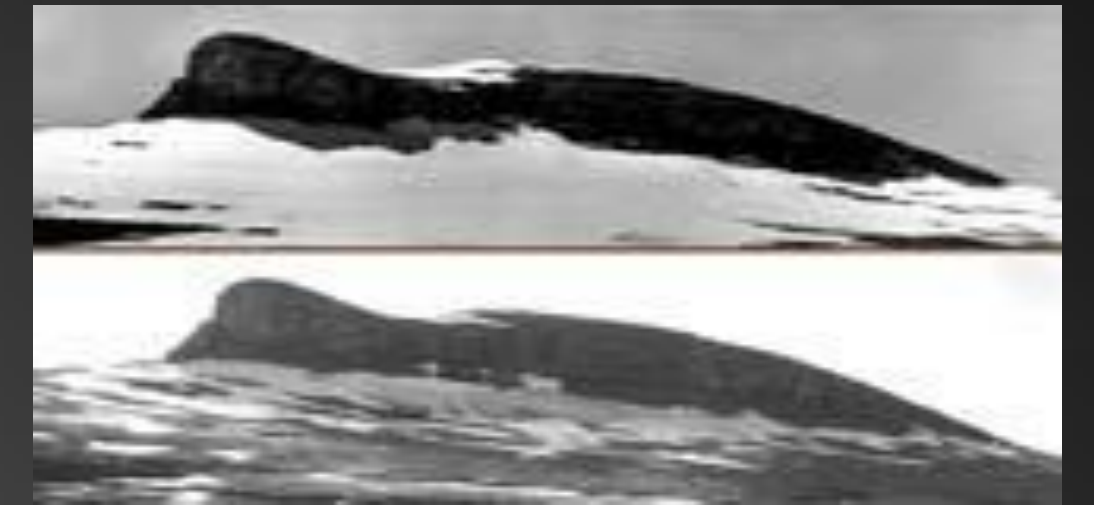


Fig. 2 Glacier Nacional Park

Abstract

In recent decades, the concept of global warming has developed increasing concern among the scientific community and general public alike. What was initially dismissed as little more than unlikely has now become a severe warning for global climate crisis threatening not only our way of life but ultimate future existence on this planet. Global warming is defined as the steady mean increase in atmospheric temperature, the primary asserted cause thereof being increased emissions and inherent atmospheric concentrations of “greenhouse gases” – carbon dioxide in particular. These gases contribute to the greenhouse effect by trapping radiation (from the sun) in the atmosphere. The largest contributor of man-made carbon dioxide is the burning of fossil fuels, especially in regards to energy production and transportation vehicles. As a result, the manufacture and utilization of cars in today’s society is not only perhaps the greatest contributor to human induced global warming but also the single major aspect that everyone can do something about. With present knowledge and considering the potential environmental influence of our collective actions, it’s the obligation of human society to restrict its climate change inducing actions if it wishes to persist and thrive as a species.

Purpose

The purpose of this research project is to increase awareness and understanding of global warming. With an emphasis on human contribution, the idea is to reflect upon our potential global environmental impact and our ability to prevent catastrophic climate change if we are indeed the underlying cause of recent temperature increases. Through clarification of global warming concepts and interpretation of statistically significant information, a greater understanding thereof is achieved. This translates to an increased ability to make informed decisions and understand what can be done to help lesson our collective impact.

Mechanism

On the molecular level, how does the greenhouse effect occur? Since Earth is surrounded by the vacuum of space, the only way energy is going to enter and/or leave the Earth’s atmosphere is via radiation. When electromagnetic radiation from the sun (or any other source of emission, for that matter) interacts with a molecule, it absorbed and thereby temporarily increases that molecule’s kinetic energy until it is transmitted via conduction, convection, or radiation thereof. The wavelength of electromagnetic radiation that will interact with any particular molecule is dependent on the properties thereof, which dictate its specific absorption signature. While some of the suns energy (in all wavelengths of visible and infrared light) is reflected by the Earth’s surface, much of it is absorbed and results in an inherent warming of the surface that is re-radiated as infrared (long wave) energy. The primary greenhouse gas culprits (water, carbon dioxide, methane, etc.) are transparent to visible light in their chemical properties yet absorb mainly infrared (thermal) radiation, enabling them to effectively retain and re-radiate energy in the form of heat. The greater the atmospheric concentration of these thermal radiation absorbing gases, the more difficult it becomes for this energy to ultimately escape the atmosphere and radiate back out into space. Much like an actual greenhouse, the heat is effectively trapped; consequently, we observe an increase in Earth’s surface temperatures. Over time, these rising surface temperatures have been demonstrated to correlate with increased greenhouse gas concentrations in the atmosphere.

Atmospheric Carbon Dioxide

While carbon dioxide isn't the only culprit responsible for global warming, it is one of the primary greenhouse gases we are responsible for emitting and second greatest (water vapor being the first) contributor to the effect. These greenhouse gases are typically measured and numerically represented in parts per million (PPM) scales, meaning the number of molecules comprising one million molecules of the solution. (The values are derived using morality and dimensional analysis.) By taking ice core samples from isolated regions, scientists are able to look back thousands to millions of years into atmospheric conditions via particles trapped and preserved in the depths of the ice. From this, we are able to determine concentrations prior to the advent of direct atmospheric measurement. The data indicates that upwards of 1,000,000 years ago, carbon dioxide concentrations remained between 190ppm-300ppm; however, around the time of the industrial revolution these values began to rapidly spike into the 300+ppm range and is fast approaching 400ppm present day.

